

A NEW ENANTIO AND DIASTEREOSELECTIVE SYNTHESIS OF 2-AZETIDINONES AS USEFUL INTERMEDIATES OF β -LACTAM ANTIBIOTICS

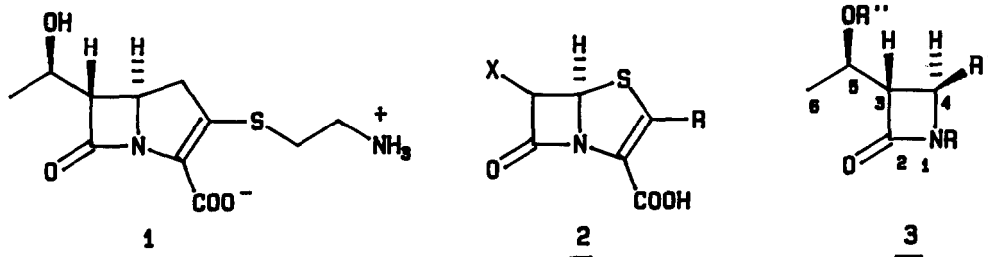
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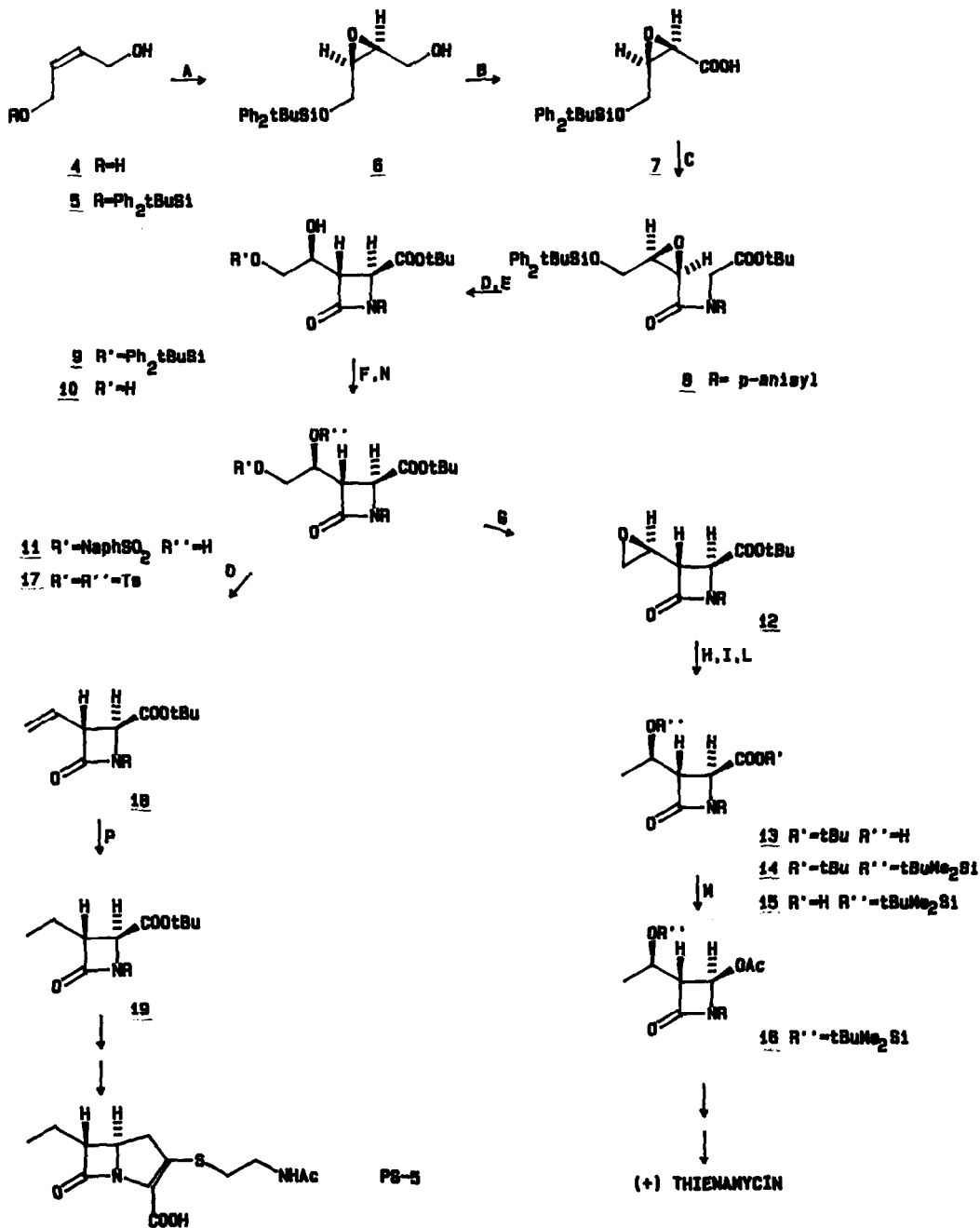
Summary: An expeditious enantio and diastereoselective route to β -lactam synthons has been developed; two novel key radical reactions allowed us to obtain two useful intermediates in the synthesis of thienamycin and PS-5.

Since 1976, after the isolation by a Merck research group¹ of (+)-thienamycin **1**, a new carbapenem with important antibacterial activity, many efficient routes to **1** and penems **2** have been developed^{2,3}.

An important intermediate common to many routes is represented by 2-azetidinone **3**, which has been obtained almost exclusively from natural precursors⁴. In this communication we wish to describe a new enantio and diastereoselective synthesis of 2-azetidinones **16** and **19**, starting from commercially available *cis*-2-buten-1,4-diol **4** (see scheme).



The monoprotected compound **5** ($t\text{BuPh}_2\text{SiCl}$, imidazole, DMF, r.t., 20 h, 30%) was epoxidized according to Sharpless⁵, to afford the chiral epoxyalcohol **6** (90%). The subsequent oxidation of **6** to **7**, was achieved in high yield buffering the reaction mixture with NaHCO_3 . Compound **7**, which is formally related to the corresponding 2,3-epoxybutane acid prepared from L-threonine^{3d}, was transformed into the epoxyamide **8** (70% from **6**). The latter was then subjected to the key cyclization step to afford β -lactam **9**: the chosen conditions (LiHMDS , THF, $T = -25^\circ\text{C}$) allowed us to obtain **9** as the only product (50% after purification by flash chromatography).



Reagents and conditions: A) Ti(O-*i*-pr)₄, L(+) DET, TBHP, CH₂Cl₂, T=-25°C (90%). B) RuCl₃, NaIO₄, NaHCO₃, H₂O, CH₃CN, CCl₄. C) p-MeO-C₆H₅-NH-CH₂COOtBu, ClCOiBu, N-methylmorpholine, 4 sieves, THF, T=-25°C (70% from **6**). D) LiHMDS, THF, T=-25°C, 15 min (50%). E) TBAF, THF, T=-20°C (90%). F) NaphSO₂Cl, Py, T=0°C (85%). G) K₂CO₃, MeOH, T=-35°C, 10h (75%). H) NaI, n-Bu₃SnH, AIBN, DME, T=75°C, 2h, (80%). I) tBuMe₂SiCl, DMAP, imidazole, r.t., 20h (90%). L) NaOH (1.1 eq), EtOH, T=50°C, 8h. M) Pb(OAc)₄, DMF, AcOH, T=70°C, 45 min (60% from **14**). N) TsCl, DMAP, CH₂Cl₂, rt, 12h (85%). O) NaI, n-Bu₃SnH, DME, AIBN, T=80°C, 24h (80%). P) Pd/C 10%, H₂ (1atm), MeOH (90%).

The correct relative stereochemistry of C-3 and C-4 was confirmed by $^1\text{H-NMR}$ decoupling. Careful deprotection of the primary hydroxyl function (T: -20°C), gave 10 in high yield (90%).

Because of the possibility of properly elaborating the hydroxy functions at C-5 and C-6, compound 10 appeared to be a new strategically important chiral synthon for the preparation of several different azetidione intermediates.

Compound 12 was prepared from 10, via the mononaphthylsulphonyl derivative 11, achieving the epoxide formation at T: -35°C^9 and then successfully transformed into 13 by a novel chemo and regioselective reductive opening of the epoxy ring (NaI, n-Bu₃SnH, DME, AIBN, T: 80°C , 2h, 80%)¹⁰ in rather good yield (80%). In order to have a chemical correlation the hydroxyl function at C-5 was protected as t-BuMe₂Si by standard procedure¹¹. Alkaline hydrolysis of the ester function (NaOH 1.1 eq, EtOH), was followed by oxidative decarboxylation of the acid 15¹², affording known 16^{13,14} (60% overall yield from 13).

On the other hand the key intermediate 10 was tosylated to 17 and, the latter, subjected to a novel radical elimination (NaI, n-Bu₃SnH, DME, AIBN, T: 80°C , 24h)¹⁵. The olefin 18, obtained as the only product of the reaction (80%) was finally hydrogenated to 19¹⁶, a potential chiral intermediate for the synthesis of the β -lactam PS-5¹⁷. The smooth obtaining of terminal olefins from 1,2 ditosylated diols by radical elimination is now investigated on several other compounds.

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References and notes

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2) for some selected references on thienamycin synthesis see: T.H. Salzman, R.W. Ratcliffe, B.G. Christensen and F.A. Bouffard *J. Am. Chem. Soc.* **102**, 6161 (1980); D.G. Melillo, I. Shinkai, T. Liu, K. Ryan and M. Sletzniger *Tetrahedron Lett.* **21**, 2783 (1980); S. Karady, J.S. Amato, R.A. Reamer and L.H. Weinstock *J. Am. Chem. Soc.* **103**, 6765 (1981).

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5) T. Katsuki and K.B. Sharpless J.Am.Chem.Soc. 102, 5974 (1980).

6) $^1\text{H-NMR}$ (60 MHz), CDCl_3 : 7.6-7.0 δ (10H, aromatic portion), 3.6 δ (2H, m, CH_2 -1), 3.4 δ (2H, m, CH_2 -4), 3.0 δ (2H, m, CH-2, CH-3), 2.5 δ (1H, t, OH), 0.9 δ (9H, s, t-Bu). The optical purity of **6** has been determined by GLC and $^{19}\text{F-NMR}$ (ee>98%) on the corresponding MPTA ester (Mosher derivative).

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8) $^1\text{H-NMR}$ (60MHz), CDCl_3 : 9.3-8.8 δ (1H, bs, -COOH), 7.6-7.0 δ (10H, m, aromatic portion), 3.8 δ (2H, d, CH_2 -4), 3.45 δ (1H, d, J=5Hz, CH-2), 3.3 δ (1H, m, CH-3), 1.0 δ (9H, s, t-Bu).

9) To avoid the formation of the eliminated product the temperature must be kept rigorously under -35°C , although with larger reaction time(10h).

10) For an extensive application of this reaction to various and different substrates see the following paper of this issue.

11) Alternatively the -OH at C-5 has been protected as THP, obtaining **13** (R^* -THP) then transformed into **16** (R^* -THP) by the described procedure. The THP group has been selectively removed in the same mild condition (AcOH, H_2O , THF, 3:3:1, T= 50°C , 1h, 90%).

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14) $^1\text{H-NMR}$ (300 MHz), CDCl_3 : 7.37-6.86 δ (4H, dd, aromatic portion), 6.66 δ (1H, s, CH-4), 4.29 δ (1H, o, CH-5), 3.79 δ (3H, s, OCH_3), 3.20 δ (1H, dd, CH-3, J=3Hz), 2.13 δ (3H, s, $-\text{OCOCH}_3$), 1.35 δ (3H, d, J=6Hz, CH_3 -6), 0.75 δ (9H, s, t-Bu), 0.1 δ (3H, s, SiCH_3), 0.08 δ (3H, s, SiCH_3). $(\alpha)_D = -67.17^\circ$ (c= 0.28, CHCl_3); lit: $(\alpha)_D = -66.86^\circ$ (13). M.p.: $70-71.5^\circ\text{C}$. HRMS: requires m/e 393.1970, found m/e 393.1970.

15) For a related method of deoxygenation of primary hydroxyl group see: Y. Ueno, C. Tanaka and M. Okawara Chem. Lett., 795 (1983).

16) $^1\text{H-NMR}$ (300 MHz), CDCl_3 : 7.20-6.80 δ (4H, dd, aromatic portion), 4.05 δ (1H, d, J=2.5 Hz, CH-4), 3.79 δ (3H, s, $-\text{OCH}_3$), 3.22 δ (1H, m, CH-3), 2.0-1.73 δ (2H, m, CH_2 -5), 1.45 δ (9H, s, t-Bu), 1.09 δ (3H, t, J=7.0Hz, CH_3 -6).

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